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Which alkene will react with br2 to give the following product

electrophiles add to the double bond of alkyls that breaks the bond π . Unlike alkyd hydrogenation, the addition of molecular or chlorine bromine to produce neighboring dialogues does not require catalysts. bromines or chlorine are the best choices for alchem alogenation. fluorine reacts very vigorously to form not only fluorurates but also different oxidation products. Adding iodine to alchenes is thermodynamically unfavorable. adding halogens (bromine or chlorine) to the double bond of alchenes could follow different mechanisms. a mechanism similar to alkyn hydrogenation would add halogen atoms on the same side and only produce the cis product. a two-phase mechanism that proceeds through an intermediate carbenium ion similar to the addition of hydrogen halogenurs with double bonds would constitute the cis and the trans product. For example, cyclopentene bromination would produce cys-1,2- cyclopentane and trans-1,2-cyclopentane. However, only the trans product is observed, indicating that the anti addition occurs selectively. Therefore the actual mechanism of alogenation is obviously different from that of hydrogenation and the addition of hx. film: addition of bromine to cyclohexane on laboratory scale. bromination and chlorination of alkyls is generally carried out in non-nucleophilic solvents such as ccl4, if the reaction is carried out in a nucleophile solvent, such as h2o or alcohol, the attacking species inThe opening of the cyclic intermediate ring is the solvent molecule rather than halogenous ion. In the case of water as a solvent, the result of the reaction is an aloidine (R2XC-C(OH)R2). In the aloidrin reaction, the reagents, HOCl or HOBr, are often generated in situ by reaction of water with chlorine or bromium, respectively. In specific reaction conditions, HOI also adds double ties. Page 2 Stereochemistry Alkenes react with bromine or chlorine to form the corresponding dialysis in an anti-addition mode. The stereochemistry of the addition of halogen is unequivocally determined by the anti addition and by the almost completely narrow rotation of the carbon-carbon bond of the halogen ion. Therefore, cys-2-butene bromination produces a race mate of (2R,3R)- and (2S,3S)-dibromobutane, while trans-2-butene bromination produces the compound of meso. Use of the illustrations below, determine whether the bromine products added to cis- and trans-butene are enantiomers or identical compounds (meso). Tab.1Products of cys-butene bromination Tab.2Products of tra-butene bromination Addition of cys-butene bromine produces an ion of achiral bromonium. The next attack of the bromide ion following the path a or b (Fig.1) gives the same product because the bromonium ion is symmetrically replaced. The product of the reaction is the competition companion of the enantiomeric products (2R,3R)-2,3-dibromobutane and (2S,3S)-2,3-dibromobutane. On the contrary, theof bromide to the trans-butene as shown in Fig. 2 produces an ion intermediate chiral bromonium that does not possess a mirror plan. However, in this special case the products (2S,3R)-2,3-dibromobutane and (2R,3S)-2,3-dibromobutane are compounds meso and therefore identical. Regardless of the fact that the reaction occurs through one or b, the same product (with a mirror plan) is always formed, meso-2,3-dibromobutane. Hit Adding halogens to alkenes using starting materials with different stereochemistry (cis or trans) always produces products with different stereochemistry. Exercise page 3 If the addition of halogens is carried out in the presence of other nucleophiles, these nucleophiles compete with halogenur ions to open the intermediate ring to three members. Etheno bromination in the presence of chloride ions produces 1,2-dibromoethan and 1-bromo-2-chloroethan, for example. Solvent Influence If the bromination or chlorination of alchenes is carried out in a nucleophilic solvent such as H2O or alcohol, the nucleophilic opening of the ring of bromonium or chloronium proceeds by attack of a solvent molecule rather than by an alogenic ion. Therefore, if water is used as a solvent the main result of alogenation is not a neighboring dialysis but a α -alogenated alcohol, aloidin. The reaction is called the reaction of the aloidine. Aloidin reactions are useful to investigate the regioselectivity of the anti addition. If they are asymmetrically replaced alchethe two C-X bonds of the halogen have different lengths. The halogen ion with the weakest and therefore the longer C-X bond to the replaced upper carbon atom is favoured vigorously, because the higher partially positive charge is more stabilized on the upper carbon atom replaced by the +I effect of alkyl substitutes and hyperconjugation. Therefore, the activation energy for the binding schism of this C-X bond is lower and the halogen ion ring is opened by the nucleophile attack of a solvent molecule on the replaced higher carbon. Exercise

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